Question Bank

ATKT EXAMINATION 2019-20

VI CBCS

Pharm. Analysis - II

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1	Which type of vibration bond angle is decrease ?A symmetricalB. AsymmetricalC. RockingD. Scissoring
2	Which type of solutions give good fluorescence ?A. DilutedB. ConcentratedC. AcidicD. Basic
3	 Which type of filter paper are mostly used in paper chromatography? A. Butter paper B. Sample paper C. Whatmann filter paper D. Filter paper
4	 Which sentence is false about UV SPECTROSCOPY ? A. Benzene is not used as solvent. B. Low temperature is give better result in uv SPECTROSCOPY. C. Chloroform is used as solvent in UV. D. UV ray wavelength is 200- 400 nm
5	Which molecules is decrease fluorescence ?A. OxygenB. HydrogenC. NitrogenD. Carbon
6	 Which material are used to prepare rod of globar source ? A. Silicon dioxide B. Silicon carbide C. Silicon oxide D. Silicon carboxamide
7	 Which material are used to made sample cell in UV-spectroscopy? A. Silver B. Silicon C. Quartz D. Aluminium

8	Which is principal of the IR spectroscopy ?A. AbsorptionB. EmissionsC. AdsorptionD. Scattering		
9	 Which is <u>not</u> use of IR spectroscopy ? A. Functional group B. Structure identify C. Polymer analysis D. Metal detecting 		
10	Which factor is responsible for the separation of the components in descending paper chromatography?A. PartitionB. AdsorptionC. GravityD. Electronegativity		
11	 Which factor does <u>not</u> affect in stationary phases in paper chromatography ? A. Thickness B. Flow rate C. Purity D. Molecular weight 		
12	 Which device is used to isolate the radiation of the desired wavelength from wavelength of the continuous spectra ? A. Monochromator B. Radiation source C. Recorder D. Collimating Systems 		
13	 Which device is used to convert light energy into electrical signal A. Amplifier B. Detector C. Sample cell D. Slit 		
14	Which detector are detected IR radiation by potential different ?A. ThermocoupleB. BolometersC. ThermistorD. Potentiometer		
15	 Which compound are used as diluent in IR sampling ? A. Potassium Bromide B. Ethanol C. Benzaldehyde D. Acetone 		
16	When there are to complex mixture are separated which development technique of Thin Layer Chromatography is used ?A. AscendingB. Two dimensionalC. DescendingD. Horizontal		

17	 When there are to complex mixture are separated which development technique of paper Chromatography is used ? A. Ascending B. Descending C. Radiation D. Two dimensional
18	When absorption intensity of compound is decreased it is calledA. Red shiftB. Blue shiftC. Hypochromic shiftD. Hyperchromic shift
19	 What is the separation principle of the paper chromatography ? A. Partition B. Adsorption C. Capilary action D. Ionization
20	 What is role of slit in uv-visible spectroscopy ? A. Monochromatic radiation to polychromatic radiation. B. Polychromatic radiation to monochromatic radiation. C. Producing narrow radiation beam D. Blocking optical path
21	Static Quenching is due to A. Neutral reaction B. Hydrogenated C. Complex formation D. Ionization
22	Rf value is A. Distance travelled by the compound at it's point of maximum. B. Distance travelled by the standard. C. Distance travelled by Solvent travelled D. Distance travelled by the compound divided by distance travelled by Solvent travelled
23	Nujol means A. Polymer B. Crude oil C. Mineral oil D. Volatile oil
24	In which vibration, one bond is one plane and other bond is another plane? A. Rocking B. Wagging C. Scissoring D. Twisting
25	In which vibration bond length is increases or decreases ? A. Rocking B. wagging C. Twisting D. Symmetrical

26	In which bending type of vibration bond angle is constant ?
	A. Scissoring B. Twisting
	C. Rocking
	D. Wagging
27	In Visible-spectroscopy Diffraction grating is consists of
2.	A. Glass
	B. Quartz
	C. Alkyl halide
	D. Silicon
28	In UV-spectroscopy Diffraction grating is consists of
	A. Glass
	B. Quartz
	C. Alkyl halide D. Silicon
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29	In photo voltaic cell which material is used as layer between cathode and anode ?
	A. Selenium
	B. Aluminum
	C. Gold
	D. Silver
30	In IR-spectroscopy Diffraction grating is consists of
	A. Glass
	B. Quartz
	C. Alkyl halide D. Silicon
31	
51	In Fluorimetric analysis, when molecules are excited in visible region which lamp is used ?
	A. Hydrogen deturium lamp
	B. Xenon lamp
	C. Tangustan lamp
	C. Mercury lamp
32	If temperature Is increased ?
	A. Fluorescence is decrease
	B. Fluorescence is increase
	C. Fluorescence remains same D. Absorbance increases but Fluorescence remains same
33	
33	If solvent is polar and electron transition is $n \rightarrow \pi^*$ which type of shift are seen in graph ?
	A. Hypsochromic shift
	B. Red shift
	C. Blue shift
	D. Hypsochromic shift with Blue shift
34	If solvent is non polar ans electron transition is $n \ge \pi$ which type
	of shift are seen in graph ?
	A. Hypsochromic shift
	B. Red shift
	C. Blue shift D. Hypsochromic shift with Red shift

35	Decrease in fluorescence intensity due to specific effect of		
	constituents of the solution itself, it is called		
	A. fluorescence		
	B. Phosphorus		
	C. Quenching		
	D. Hydration		
36	Collimator is used for?		
	A. EMR To focus on of the entrance slit.		
	B. EMR To get narrow beam source		
	C. EMR To fall on sample cell		
	D. EMR To disperse radiation		
37	is required for the IR spectroscopy ?		
	A. Dipole moment		
	B. Spin moment		
	C. Round moment		
	D. Linear moment		
38	is responsible for uv absorption in a compound?		
	A. Auxochrome		
	B. Chromophore		
	C. Colour		
	D. Sigma bonds		
39	IF Molecular weight is increase then what happen with		
	fluorescence		
	A. fluorescence increase		
	B. No effect on fluorescence		
	C. fluorescence increases with increase in absorbance		
	D. fluorescence decrease		

ANSWER KEY

1	D	26	С
2	A	27	А
3	С	28	В
4	С	29	А
5	A	30	С
6	В	31	В
7	С	32	В
8	A	33	D
9	D	34	А
10	A	35	С
11	D	36	А
12	A	37	А
13	В	38	В
14	A	39	D
15	A		
16	В		
17	D		
18	С		
19	A		
20	С		
21	С		
22	D		
23	С		
24	D		
25	D		